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Reviews

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RareBooksClub. Paperback. Book Condition: New. This item is printed on demand. Paperback. 38 pages. Original publisher: Aberdeen Proving Ground, MD : Army Research Laboratory, 2010 OCLC Number: (OCoLC)710812456 Subject: Nitric acid. Excerpt: . . . Appendix. Mechanism Data Tables The monomethylhydrazine-red fuming nitric acid (MMH-RFNA) mechanism discussed in this report comprises thermochemical parameters and rate expressions for 81 species and 513 elementary reactions. Thermochemical parameters are given in table A-1, and the rate expressions are given in table A-2. The thermochemical parameters include (1) heat of formation at 298 K $H^{\circ}(298)$, (2) entropy at 298 K $S^{\circ}(298)$, and (3) the constant pressure heat capacity $C_p(T)$ as a function of temperature T . For reactions whose rates (k) are not a function of pressure, the Arrhenius expression $k = A \exp(-E/RT)$ calculates the rate at a specified temperature from the given frequency factor (A), temperature exponent (b), activation energy (E), and the ideal gas constant (R). For reactions involving a generalized collider species (M) (and whose rates are therefore pressure dependent), an effective collider concentration (C) is calculated from the expression $C = \sum_i p_i (RT)^{X_i}$, where p is the pressure, X is the mole fraction, and i is the collider efficiency of species i . Collider efficiencies are assumed to be 1.0 unless otherwise specified. Then, one of three expressions is employed for the rate calculation. If T and $H^{\circ}(S)$ is specified, the 1 Tsang and Herron form is used and constants for that form are given,...



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